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# Size-selected cluster deposition in liquid environments: Pd<sub>12</sub> on Au films

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#### Motivation

- Metal clusters with atomic size-selection promise material-efficient solutions to catalytic challenges such as water splitting or C-C coupling reactions
- Taking advantage of their high surface area and size-specific properties
- Preparation of stable and active clusters (< 100 atoms) at the solid/liquid interface demands new approaches
- Exploit intrinsic properties of POMs: Distinct number of metal atoms
- Huge variarity in kind of metal and atomic numbers of usable precursors



[SrPd<sub>12</sub>O<sub>6</sub>(OH)<sub>3</sub>(PhAsO<sub>3</sub>)<sub>6</sub>(OAc)<sub>3</sub>]<sup>4-</sup> Polyoxometalate (POM) with a discrete number of 12 Pd atoms are reduced at the solid/liquid interface. [1]

#### Novel preparation approach of metal clusters

• POM show no specific adsorption, but change in PZC

## Pd<sub>12</sub> clusters on Au(111)

- Immersing flame-annealed Au(111) crystal in Palladium(II)-oxo solution
- 25 x 200 ms reduction pulses to 0 V vs RHE
- Particle analysis suggests Pd<sub>12</sub>, Pd<sub>24</sub>, Pd<sub>36</sub>, Pd<sub>48</sub>, Pd<sub>60</sub> cluster sizes, area in good agreement with UHV studies [3]
- Amount of pulses influences cluster coverage of surface, exact dependence still unclear

60

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No prefered deposition site of clusters encountered



- POM Pd (II) gets reduced on Au at 0.2 V vs RHE
- Concomitant hydrogen evolution on deposited Pd
- Evolved hydrogen reduces POM as well
- Pulsed deposition approach to exclude H<sub>2</sub> evolution
- 200 ms pulses to 0 V vs RHE result in monodisperse Pd clusters



First and second cycle of Au in  $H_{3}PO_{4}/NaOH$  with  $Pd_{12}$ -POM. Deposition of Pd occurs below 0.2 V vs RHE.

#### Local measurement - Electrochemical STM

Combines local topographic information with integral electrochemistry





Pd deposition on Au(111): STM with  $U_{tun} = 72 \text{ mV}$ ,  $I_{tun} = 0.9 \text{ nA}$ . Detected particles show multiples of Pd<sub>12</sub> clusters. Coverages depends on amount of reductive pulses.

### Summary and Outlook

Acknowledgements

We show the successful deposition of Pd<sub>n</sub> clusters via POM decomposition on Au(111) by means of a potential pulse. The well-defined cluster sizes promise a good understanding of the non-scalable properties.

Further studies will concern the possible ripening processes and reactivity by appropriate choice of POM precursors.

• Tip as fourth electrode probing tunnel current

• Investigation of ripening processes, reactivity measurements via noise level [2]



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#### References

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