# Known Errors and Typos in "concepts of Modern Catalysis and Kinetics" by

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We apologies for any inconvenience these errors and typos has caused.

## Chapter 2

Page 26 eq. 8 should have read

$$r = k^{+} [A]^{v_a} [B]^{v_b} - k^{-} [C]^{v_c} [D]^{v_d}$$

Page 30 line 11 from top equation should read

$$\Delta G^{\circ} = \Delta H^{\circ} - T \Delta S^{\circ}$$

Page 36 eq. 45 should read

$$k(T) = ve^{-E_a/RT}$$

## Chapter 3

**Page 94 eq. 66** should  $\delta$  be replaced by  $\partial$  so that it reads:

$$\mu_{i} = -k_{\mathrm{B}}T \left[ \frac{\partial \ln(Q_{i})}{\partial N_{i}} \right] = -k_{\mathrm{B}}T \left[ \frac{\partial \ln\left(\frac{q_{i}^{N_{i}}}{N_{i}!}\right)}{\partial N_{i}} \right] \approx -k_{\mathrm{B}}T \ln\left(\frac{q_{i}}{N_{i}}\right)$$

Page 96 same error as above so that eq. 76 reads:

$$\mu_{i} = -kT \left[ \frac{\partial \ln(Q_{i})}{\partial N_{i}} \right] = -k_{B}T \left[ \frac{\partial \ln\left(\frac{q_{i}^{N_{i}}}{N_{i}}!\right)}{\partial N_{i}} \right] \approx -k_{B}T \ln\left(\frac{q_{i}}{N_{i}}\right)$$

Page 96-97 in eq. 76, 78, and 80 k should read  $k_B$ 

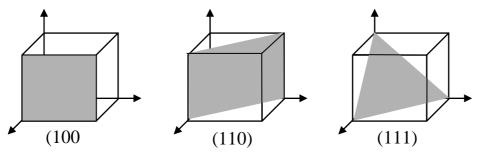
And equation 81 should read 
$$\frac{q^{trans}}{V} = \frac{\left(2\pi m_x k_B T\right)^{3/2}}{h^3}$$

Page 99 in eq. 89-91: R should be replaced with k<sub>B</sub>

Page 100 table 3.4 1 line  $K_{H_2}(T)$   $p_H/p_0$  instead of  $K_{H_2}H/p_0$   $p_w/p_0$ 

# Chapter 5

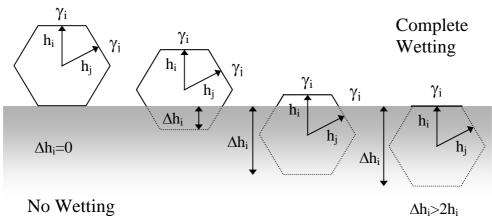
Page 169 Figure 5.2 should look like this:



Page 173 8<sup>th</sup> line from bottom should refer to fig. 5.5 and not 5.6.

**Page 180 1. line** under figure 5.13 should read "The surface energy  $\gamma(ijk)$  for each surface  $\gamma(ijk)$ ..."

**Page 181 Figure 5.15**  $\Delta h_i$  should replace  $\Delta h_j$  so it look like this



Page 185 eq. 10 should read 
$$\theta_0 = \left[1 + \frac{k_2}{k_1} \frac{\left(\frac{F}{k_2}\right)}{1 - \left(\frac{F}{k_2}\right)}\right]^{-1}$$

Page 186 4th line from top: N<sub>0</sub>=PV<sub>0</sub>/k<sub>B</sub>T i.e. replace R with k<sub>B</sub>

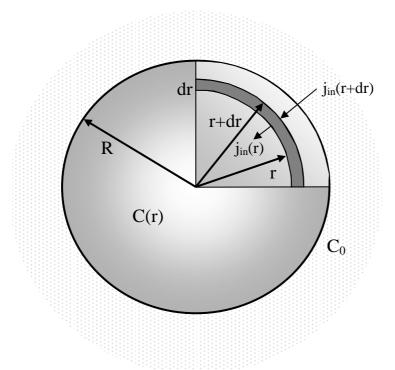
#### Page 204 item 4 should read:

4. Establish ideal flow patterns: This is usually assumed to be the case for plugflow (PFR) and continuously stirred tank reactors (CSR), but are all conditions for ideal mixing fulfilled? For example, a rule of thumb is that the diameter d of the PFR should be at least  $10 \times$  the diameter  $d_p$  of the catalyst particles to eliminate the influence of the reactor wall. Also, the amount of catalyst should be sufficient to avoid axial gradients. Another rule is that the length of the bed L should be  $50 \times$  the particle diameter, i.e.  $L > 50d_p$ . Higher values are preferable, but these may cause other problems such as temperature gradients and pressure drops.

#### Page 207 eq. 18 should read:

$$\operatorname{rate}\left[\frac{\operatorname{mol}}{\operatorname{s}}\right] = V\left[\operatorname{m}_{\operatorname{cat}}^{3}\right] S\left[\frac{m_{\operatorname{surfcat}}^{2}}{m_{\operatorname{cat}}^{3}}\right] k\left[\frac{m^{3n}}{m_{\operatorname{surfcat}}^{2} \operatorname{mol}^{n-1} \operatorname{s}}\right] C^{n}(r,t) \left[\frac{\operatorname{mol}^{n}}{\operatorname{m}^{3n}}\right]$$
(18)

Page 207 figure 5.33 should have looked like this:



**Page 209 eq. 28** should look like this 
$$\Phi_S = R \sqrt{\frac{Sk}{D_{\text{eff}}}}$$
  $\omega = \frac{\Phi_s}{R} = \sqrt{\frac{Sk}{D_{\text{eff}}}}$ 

## Page 211 eq 36 and 38 should read

$$\begin{split} E^{\text{app}} &= RT^2 \frac{\partial \ln(r)}{\partial T} = \Delta E_{\text{act}} \\ E^{\text{app}} &= RT^2 \frac{\partial \ln(\text{Rate}_{\text{diff}})}{\partial T} = \frac{\Delta E_{\text{act}}}{2} + \frac{\Delta E_{\text{diff}}}{2} \approx \frac{\Delta E_{\text{act}}}{2} \quad \text{when} \quad \frac{\Delta E_{\text{diff}}}{2} \to 0 \end{split}$$

#### Page 213 ed. 46 should read

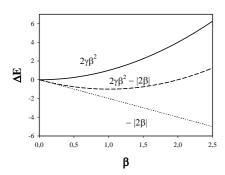
$$E^{\rm app} = RT^2 \frac{\partial \ln(r_{\rm diff})}{\partial T} = \frac{\Delta E_{\rm act}}{2} + \frac{\Delta E_{\rm diff}}{2} \approx \frac{\Delta E_{\rm act}}{2} \quad \text{when} \quad \frac{\Delta E_{\rm diff}}{2} \to 0$$

Page 212 eq. 42 last parenthesis should be deleted so it reads:

$$C(x) = C_0 \frac{\cosh\left[\omega(L-x)\right]}{\cosh(\omega L)} = C_0 \frac{\cosh\left[\Phi_s \frac{(L-x)}{L}\right]}{\cosh(\Phi_p)}$$

# Chapter 6

Page 221 figure 6.5 miss a straight line so it should look like this:



**Page 236 eq. 53** middel replace  $\int \Psi_a H \Psi_k$  with  $\int \Psi_k H \Psi_k$ 

**Page 238 eq. 66, 67 and 71** should  $\varepsilon_a$  be replaced by  $\varepsilon_k$ . This has the consequence that the  $\varepsilon_a$  under the summation in eq 68, 69, and 71 should be relplaced by  $\varepsilon_k$ .

**Page 238** The line in between eq. 66 and 67 should read: allowing us to solve for  $G_{ka}$ :

**Page 247 eq. 77** should read  $\delta = (\varepsilon_a - \varepsilon_d) > 0$ 

Page 277 eq. 80 should read

$$E_{\text{d-hyb}} = -(1-f)\frac{2\beta^2}{\delta} + 2(1+f)\gamma\beta^2$$

$$= -(1-f)\frac{2\beta^2}{(\varepsilon_a - \varepsilon_d)} + 2(1+f)\gamma\beta^2$$
(80)

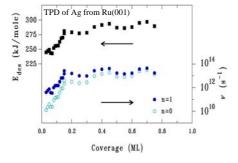
**Page 251 eq. 85** in the 3th term  $2(1+f)\gamma_{2\pi}\beta_{2\pi}^2$  should read  $2(1+f)\gamma_{5\sigma}\beta_{5\sigma}^2$ .

# **Chapter 7**

Page 268 eq. 8 the term "lim its" should simply be removed twice.

Page 277 5<sup>th</sup> and 6<sup>th</sup> line from top the reference to Eq. (12) should be Eq. (14).

Page 278 figure 7.8 right axis should read like shown below



Page 291 eq 38 should read

$$NH_2 *+ H * \longleftrightarrow NH_3 *+ * \frac{d\theta_{NH_3}}{dt} = k_5^+ \theta_{NH_2} \theta_H - k_5^- \theta_{NH_3} \theta_* = 0 \Rightarrow \theta_{NH_2} = \frac{\theta_{NH_3} \theta_*}{K_5 \theta_H}$$

And eq. 39 should read:

$$NH_3 * \longleftrightarrow NH_3 + * \frac{d\theta_{NH_3}}{dt} = -k_6^+\theta_{NH_3} + k_6^-P_{NH_3}\theta_* = 0 \Rightarrow \theta_{NH_3} = \frac{1}{K_6}P_{NH_3}\theta_*$$

## Chapter 8

Page 340 electric power consumption was  $4.6 \times 10^{19} \,\mathrm{J} \,\mathrm{year}^{-1}$  not  $3.8 \times 10^{19} \,\mathrm{J} \,\mathrm{year}^{-1}$ 

**Page 345 table 8.9** the density of lig.  $H_2$  is 0.0708 not 0.078.

## **Questions and Exercises.**

Page 417 eq.2 in exercise 2.10 should read: (2)  $CO+* \longrightarrow CO*$ 

**Page 418 top second line** should read:  $S_0^{\text{H}_2} = S_0^{\text{CO}} = ...$ 

Page 418 elementary step 3 and 4 in exercise 2.11 should read

(3) 
$$CO_2* + H* \longrightarrow HCOO* +*$$

**Page 429 question 4 in exercise 5.**1 the standard conditions is at 298K not 273 as written.

Page 430 second last line of exercise 5.2 replace reproduce with reduce.

Page 431 6<sup>th</sup> line should start with 7)

**Page 432 2. line** from top should read  $M_W=58.71g$ .

Page 432 in 7.th line dosage of 0.2 bar s.

Page 432 the x-axis should be in units of bar\*sec not torr\*sec

**Page 437 line 2** 
$$NO^*+^* \to N^*+O^*$$
 *RLS*

**Page 437** line 6 
$$N^* + N^* \in N_2^* + N_2^*$$

**Page 440 Eq. 3** 
$$O_2 * + H^* \rightarrow HO_2 * + * RLS$$

## Appendix A

**Page 443 line 9** Electron mass  $m_e = 9.110 * 10^{-31} kg$ 

Page 443 line 16 Vacuum Permitivity 8.854x10<sup>-12</sup>Cm<sup>-1</sup>V<sup>-1</sup>

**Page 444 2. line** 1 bar =0.98692atm=750.06 mmHg=100000Pa=100000kgm<sup>-1</sup>s<sup>-2</sup>